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A Current Review on Analytical Tools for Determination of New Oral Antidiabetic Drugs, Empagliflozin, Linagliptin and Biguanides in Bulk Materials, Pharmaceuticals & Biological Samples

S. Baokar^{1,2*}, K. Mane¹, A. Bhujbal¹, S. Pharande¹, G. Patil¹, R. Patil¹, P. Jain³ and A. Pandey⁴

¹Delonix Society's Baramati College of Pharmacy, Barhanpur, Baramati Mah. 413102, India. ²Mansarovar Global University, Bhopal (M. P.) 466111, India. ³Scan Rrsearch Laboratories, Bhopal (M. P.) 466111, India. ⁴Mansarovar Global University, Bhopal (M. P.) 46611, India.

Authors' contributions

This work was carried out in collaboration among all authors. Author SB designed the study, performed the statistical analysis, wrote the protocol and wrote the first draft of the manuscript. Authors KM, SP, AB, GP, RP and PJ managed the analyses of the study. Author AP managed the literature searches. All authors read and approved the final manuscript.

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Review Article

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ABSTRACT

Worldwide the R & D divisions of Pharma industry are actively involved in the development of new therapeutic agents. These agents may be either new entities or partial structural modification of the existing one. The recent FDA statistics represent that the average number of drug filings are increasing every year in the thrust areas like anti-cancer agents, anti-diabetic, antibiotics, cardio-vascular drugs, respiratory drugs etc. Sodium glucose co-transporter-2(SGLT-2) inhibitors, dipeptidyl peptidase-4 (DPP-4) inhibitors and biguanides are effective oral anti-diabetic agents used in treatment of type 2 Diabetes Mellitus. Therefore, the necessity to explore and compare the existing analytical and bioanalytical assays used for determination of such drugs either single or in

^{*}Corresponding author: E-mail: krishnabaokar@gmail.com;

combination is crucial. Many methods were reported in the literature for the bio-analysis and analysis of four novel gliptins combinations, empagliflozin-linagliptin, empagliflozin-metformin HCl, linagliptin-metformin HCl, empagliflozin-linagliptin-metformin HCl combination with application on Glyxambi®, Synjardy®, Jentadueto®, Trijardy® XR tablets respectively. Furthermore, this review offered an overview of different methods used for determination of every drug alone as empagliflozin from SGLT-2 inhibitors, linagliptin from DPP-4 inhibitors and metformin from biguanides in a tabulated comparative way. Moreover, the current review emphasizes the most common stability indicating assays to be of interest to the analysts in the area of drug control. This review helps in understanding the further need for the development of analytical methods for the estimation of such drugs.

Keywords: Anti-diabetic; analytical methods; empagliflozin; linagliptin; metformin HCl.

1. INTRODUCTION

Analysis is vital in any product or service, and it is also important in drug because it involves life Analytical method development [1]. and analysis validation for therapeutic the components and associated substances play an important role in the discovery, development and manufacture of pharmaceuticals and natural medicinal compounds. Analytical instruments play a major role in the process to achieve high quality and reliable analytical data. Thus everyone in the analytical laboratory should be concerned about the quality assurance of equipment. Analytical method could be spectral, chromatographic, electrochemical, hyphenated miscellaneous. Analytical method or development is the process of selecting an accurate assay procedure to determine the composition of a formulation. It is the process of proving that an analytical method is acceptable in laboratory for use to measure the concentration of subsequent samples. Analytical methods should be used within GMP and GLP environments and must be developed using the protocols and acceptance criteria set out in the ICH guidelines Q2(R1) [2-6]. Diabetes mellitus is one of the chronic metabolic disorders that have reached to epidemic proportions worldwide [6]. It may be defined as a group of physiological impairments or dysfunctions distinguished by chronic hyperglycemia as a consequence of the inability of cells to use glucose for essential biological processes. The underlying causes of diabetes mellitus include insulin insensitivity, impaired insulin secretion or excessive glucagon secretion [7]. In addition to diabetes mellitus being responsible for an increased rate of mortality and morbidity incidences in patients [6], it is usually also associated with a number of different damaging effects on various tissue. cellular and even organ functions. Some of the complications of diabetes mellitus include

hypertension [7], cardiovascular diseases [8], diabetic retinopathy [9], renal nephropathy [10] etc. Type 1 (Insulin dependent Diabetes Mellitus-IDDM), occur mostly in juvenile and when secretion of insulin is diminished. Management of type 1 DM is achieved through intake of exogenous insulin. Type 2 (Non-insulin dependent Diabetes Mellitus - NIDDM), is more common in older adults however its incidence among teenagers have boosted in the current years mainly due to unhealthy lifestyle. The clinical management as well as prevention of diabetes requires superlative control of factors like blood glucose, blood pressure, lipid concentrations, body weight etc. that have the potential to cause complications. This can be accomplished by maintaining a strict diet regime and regular exercise or by the use of antidiabetic medications or a combination therapy including both [7]. Oral anti-diabetic drugs are initiated in case of type 2 DM that had inadequate response toward lifestyle change including calorie restriction and increase in physical activity. SGLT-2 inhibitors, DPP-4 inhibitors and biguanides are effective oral antidiabetic agents used in treatment of type 2 Diabetes Mellitus. Therefore, the necessity to explore and compare the existing analytical and bioanalytical assays used for determination of such drugs either single or in combination is crucial. Diverse techniques like spectrophotometry, capillary electrophoresis (CE), high-performance liquid chromatography (HPLC), liquid chromatography--mass spectrometry (LC-MS) and high performance thin layer chromatography (HPTLC) have been used for the analysis of different formulations. These techniques offer a sensitive and rapid determination of these 3 drugs in bulk drug, pharmaceutical formulations, in biological matrices and could be used for quality control assay and pharmacokinetic assays. In the present review we have compiled the published analytical methods reported for the determination of empagliflozin, linagliptin, metformin HCl in bulk drug, various pharmaceutical formulations and in biological samples.

1.1 Oral Hypoglycemic

Oral hypoglycemic are chemical agents, which are used to control the blood glucose level of diabetes patients. These drugs are also called anti-hyperglycemic agents that act by increasing insulin secretion, promoting organs sensitivity towards insulin and sometimes reducing the absorption of glucose from the gastrointestinal tract into the blood [11].

1.2 Biguanides

The discovery of biguanide and its derivatives in the management of diabetes started in the middle ages. Galega officinalis, an herbaceous plant was found to contain guanidine, galegine and biguanide, which was found to decrease blood glucose levels [12]. Biguanides are widely prescribed antihyperglycemic agents that suppress hepatic glucose production, increase peripheral glucose uptake and moderately reduce LDL cholesterol and triglyceride levels. Glucose control with the aid of biguanides appears to decrease the risk of diabetes-related complications, and is not associated with weight gain. The most common adverse effect of biguanides is gastrointestinal distress, including diarrhoea, cramps, nausea, vomiting, and increased flatulence. Long-term use of biguanides has been associated with decreased absorption of vitamin B12 [13].

1.3 Metformin

Metformin (MET Fig. 1A) chemically 3-(diaminomethylidene)-1, 1-dimethylguanidine, lowers plasma glucose level via four distinct mechanisms: inhibits hepatic gluconeogenesis glycogenolysis, stimulates peripheral and glucose uptake, utilization and increases insulin sensitivity in and delays/inhibits intestinal glucose absorption stimulates intracellular glycogen synthesis via glycogen synthase and increases membrane glucose transporters capacity [14]. MET is used basically in case of obese patients to increase fatty acid oxidation, reduce glucose absorption rate from GIT as well as, increase glucose uptake by phosphorylating GLUTenhancer factor [15,16]. Moreover, a research revealed that MET actives AMP-activated protein

kinase that promotes gluconeogenic genes in liver [17]. MET is used as monotherapy, but it can be used as a combination with sulfonylureas agents. However, if the patients has renal failure syndrome, it may lead to lactic acidosis [18].

1.4 Dipeptidyl Peptidase-4 (DPP-4) Inhibitors

Dipeptidyl peptidase-IV (DPP-4) inhibitors are a class of oral antidiabetic drugs that considered as oral alternatives to the GLP-1 agonist. They increase circulating concentrations of the incretin gastrointestinal hormones: GLP-1 and GIP polypeptides, by preventing their degradation by DPP-4 enzyme. This enhances insulin secretion from pancreatic β -cells, lowers glucose levels in an insulin-dependent manner and reduces plasma glucagon secretion from pancreatic qcells. Thus, reducing endogenous glucose production [19,20]. The most common used DPP-4 inhibitors are: saxadiptin, vildadiptin, sitagliptin, linagliptin and alogliptin. Their effectiveness is comparable with the other hypoglycemic agents present in current times; even sometimes, these drugs are more effective with very little chances of hypoglycemia [21].

1.5 Linagliptin

Linagliptin (LIN Fig. 1B) chemically 1H-Purine-2, 8-[(3R)-3-amino-1-piper-idinyl]-7-(2-6-dione, butyn-1- yl)-3, 7-dihydro-3-methyl-1-[(4-methyl-2quinazolinyl) methyl] is a DPP-4 inhibitors in the treatment of type-II diabetes. DPP-4 is an enzyme that degrades the hormones, glucagon like peptides -1 (GLP-1) and glucose dependent insulin tropic polypeptide (GIP).Both GLP-1 & GIP increases the insulin biosynthesis and secretion from pancreatic α -Cells resulting in the reduction of hepatic glucose output [22]. Amongst all the members of DPP peptidase family, linagliptin is approximately 40,000 fold more selective towards the enzyme DPP-4. After binding, its dissociation rate from the active site of DPP-4 enzyme is also slow owing to its sustained release. Linagliptin mediates competitive DPP4 inhibition which is reversible at the same time [23,24].

1.6 Sodium Glucose Co-Transporter 2 Inhibitors

Sodium-glucose co-transporter 2 (SGLT-2) inhibitors, also known as gliflozins, are a novel class of pharmacologic agents developed for the management of T2DM. They play an important

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role in alucose homeostasis by filtering and reabsorbing glucose in the proximal tubules. The mechanism of action of SGLT-2 inhibitors is independent of β-cell function and has shown favorable effects on blood pressure, body weight and albuminuria [25,26]. These benefits have considerable resulted in reduction in cardiovascular and renal events in patients with T2DM as evident from several cardiovascular outcome trials [27-30]. Currently there are four SGLT-2 inhibitors namely dapagliflozin (Farxiga), empagliflozin (Jardiance), canagliflozin (Invokana) and ertugliflozin (Steglatro) that are approved by the FDA and are widely studied in large cardiovascular outcome trials [27-30]. Besides, there are others such as ipragliflozin, luseogliflozin, janagliflozin. sotagliflozin, remogliflozin and tofogliflozin which are at different stages of clinical investigation leading to possible approval. Due to their unique mechanism of action, SGLT-2 inhibitors have the potential for use in combination therapy with other glucose lowering agents. They have shown clinically meaningful reductions in glycated hemoglobin, body weight and systolic blood pressure when given as add-on or fixed-dose combination therapies such as Invokamet (canagliflozin-metformin), Xigduo XR (dapagliflozin-metformin), Qtern (dapagliflozinsaxagliptin), Synjardy XR (empagliflozin-(empagliflozinmetformin) and Glyxambi linagliptin) for patients with T2DM [25,26].

1.7 Empagliflozin

Empagliflozin (EMP Fig. 1C) chemically (2S,3R,4R,5S,6R)-2-[4-chloro-3- [[4-[(3S)-oxolan-3-yl]oxyphenyl]methyl]phenyl]-6- (hydroxymethyl)oxane-3,4,5-triol.

SGLT2 is the predominant transporter responsible for reabsorption of glucose from the glomerular filtrate back into the circulation. EMP is an inhibitor of SGLT2 reduces renal reabsorption of filtered glucose and lowers its renal threshold and thereby increases urinary glucose excretion. EMP increases urinary excretion of glucose by markedly reducing the renal tubular threshold for glycosuria. This leads to excretion of 60-100 g/day of glucose, improving glucose control with low risk of hypoglycemia and results in loss of 240-400 kCal/ day into the urine with associated weight reduction. In addition, a decrease in blood pressure is seen due to osmotic diuresis of glucose and natriuresis of co-transported sodium [31,32].

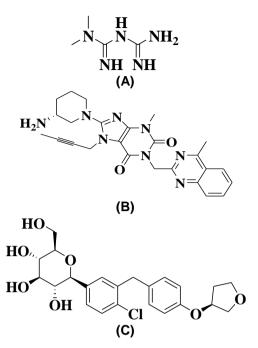


Fig. 1. Chemical structure of (A) Metformin, (B) Linagliptin, (C) Empagliflozin,

2. ANALYTICAL METHODS

2.1 HPLC/UPLC Method

Determination of MET, EMP and LIN alone or in combination with other drugs in biological samples as well as pharmaceutical formulations carried out by HPLC/UPLC are listed in Table 1 [33-61].

2.2 Spectrophotometric Method

Various spectrophotometric methods used in MET, EMP and LIN in pharmaceutical formulations and biological samples are presented in Table 2 [62-74].

2.3 HPTLC Method

MET, EMP and LIN estimation in various pharmaceutical dosage forms was successfully carried out using HPTLC methods and is summarized in Table 3 [75-80].

2.4 Stability-Indicating and LC-MS/MS Method

Stability-indicating HPLC and UPLC methods for determination of MET, EMP and LIN in pharmaceutical formulation were given in Table 4 [81-99] and LC-MS/MS method for determination of MET, EMP and LIN in biological matrix and pharmaceutical dosage form was given in Table 5 [100-110].

Sr. no	Name of Drug/ Formulation/ Biological Fluid	Stationary Phase	Mobile Phase	Detection wavelength (Nm)	Ref.
1	EMP-Tablet	Hypersil BDS 150mm X 4.6 Mm, 5μ	0.1% OPA: Acetonitrile (70:30% V/V)	233 nm	[33]
2	EMP-Tablet	Enable C ₁₈ G (250 X 4.6 Mm,5µ)	Methanol: Water (70:30 % V/V)	233 Nm	[34]
3	EMP-Tablet	Shim Pack C ₁₈ (250 Mm×4.6 Mm,5 µm)	Acetonitrile: Water (60:40 V/V)	223 Nm	[35]
4	EMP-Plasma	Acquity UPLC BEH C ₁₈ (50 Mm×2.1 Mm,1.7 μm)	Trifuoroacetic Acid (0.1%, Ph 2.5): Acetonitrile (60:40 V/V)	210 Nm	[36]
5	EMP-LIN-Synthetic Mixture	Agilent C ₁₈ (4.6×150mm) 5µ	Methanol: Phosphate Buffer (KH ₂ PO ₄ And K ₂ HPO ₄) Ph 3 (70:30 V/V)	254 Nm	[37]
6	EMP-LIN-Tablet	Agilent C ₁₈ (4.6×150mm) 5µ	Methanol: Phosphate Buffer (KH_2PO_4 And K_2HPO_4) Ph 3 (70:30 V/V)	254 Nm	[38]
7	EMP-LIN-Plasma	Discovery C ₁₈ (250×4.6×5)	Buffer: Acetonitrile (68:32 V/V)	218 Nm	[39]
8	EMP-LIN-Tablet	Thermo C_{18} (250mmx4.6mm, 5µm)	Phosphate Buffer Ph 3.4: Methanol (70:30 V/V)	240 Nm	[40]
9	EMP-LIN-Tablet	Equisil BDS C ₁₈ (4.6 × 250 Mm) 5µ	Methanol: Water (40:60 V/V)	224 Nm	[41]
10	EMP-LIN-Tablet	Youglin C ₁₈ (250 Mm X 4.6 Mm) 5 µm	Methanol: Water (80:20 V/V)	233 Nm	[42]
11	EMP-LIN-Tablet	Kromosil (250 X 4.6 Mm, 55 μm)	Buffer Ph 4.8: Acetonitrile (70:30 V/V)	286 Nm	[43]
12	EMP-LIN-Tablet	Kromasil 250 X 4.6 Mm, 5mm	0.1% O-Phosphoric Acid Buffer: Acetonitrile (60:40%V/V)	230 Nm	[44]
13	EMP-MET-Tablet	C ₁₈ (150 Mm X 4.6 Mm;5µ)	Methanol: Phosphate Buffer (40:60 V/V)	255 Nm	[45]
14	EMP-MET- Canagliflozin- Dapagliflozin- Tablet	Inertsil ODS C ₁₈ (250×4.6 Mm-5 μ m)	Acetonitrile: 0.05 M Phosphate Buffer Ph 4 (65:35 V/V)	212 Nm	[46]
15	EMP-MET- Canagliflozin- Dapagliflozin- Tablet	Prontosil (Lichrosorb 100-5- NH ₂)	Nah ₂ po ₄ Buffer (10 Mm, Ph 2.8):Acetonitrile (18.5:81.5, V/V),	225 Nm	[47]
16	EMP-MET-Tablet	Symmetry C_{18} (4.6×150mm) 5µ	Methanol: Phosphate Buffer (KH_2PO_4 And K_2HPO_4) Ph 3(70:30 V/V)	240 Nm	[48]
17	EMP-MET-Tablet	X-Select-HSS C ₁₈ SB (4.6-Mm X 25-Cm; 5- µm)	Phosphate Buffer : Acetonitrile (60:40 V/V)	255 Nm	[49]

Table 1. RP-HPLC/UPLC methods for determination of MET, EMP and LIN

Sr. no	Name of Drug/ Formulation/ Biological Fluid	Stationary Phase	Mobile Phase	Detection wavelength (Nm)	Ref.
18	EMP-MET-LIN-Tablet	Acclaim TM RSLC 120 C ₁₈ (100 Mm × 2.1 Mm, 2.2 μm)	Phosphate Buffer Ph 4: Methanol (50:50, V/V)	225 Nm	[50]
19	LIN-MET-Plasma	Grace Vyadyec Genesis CN (150 × 4.6 Mm, 4 μm)	Acetonitrile: 0.01M Phosphate Buffer Ph 7.0 (75:25 V/V)	237 Nm	[51]
20	LIN	Chiralpak AD-H (250*4.6 Mm*5 µm)	Ethanol: Methanol : Diethylamine (90:10:0.1 V/V/V)	225 Nm	[52]
21	EMP-MET-LIN-Saxagliptin- Teneligliptin-Pioglitazone- Dapagliflozin- Gliclazide- Tablet	Waters Reliant™ (250 Mm × 4.6 Mm, 5 µm).	Acetonitrile: 0/01% V/V Formic Acid Buffer	-	[53]
22	LIN-MET-Plasma	Lichrosphere 100 RP 18e (125 Mm × 4.0 Mm, 5 Mm)	Potassium Dihydrogen Orthophosphate (0.05 M, Ph 4.6): Methanol (70:30 V/V)	267 Nm	[54]
23	MET- Sitagliptin-Tablet	Symmetry C ₁₈ (100 Mm × 2.1 Mm, 2.2 Mm)	(Methanol 20%), Ph 3.5	220 Nm	[55]
24	MET-Teneligliptin- Tablet	Cosmosil C ₁₈ , 250X4.6mm, 5µm	Methanol: Water Ph 3.5 (50:50 V/V)	242 Nm	[56]
25	MET-Sitagliptin- Vildagliptin	Symmetry® Cyanide (150 Mm × 4.6 Mm, 5 Mm)	Phosphate Buffer Ph4.6: Acetonitrile (30:70 V/V)	210 Nm	[57]
26	MET-Gliclazide-Tablet	Alltima CN (250 Mm 4.6 Mm X5µm)	20 Mm Ammonium Formate Buffer Ph 3.5: Acetonitrile (45:55 V/V)	227 Nm	[58]
27	MET- LIN-Plasma	Onyx C_{18} Monolithic (100mm× 5µm)	CAN: Meoh: 0.01% HCOOH Ph 3 (30:13.59:56.41 V/V)	220 Nm	[59]
28	MET- Canagliflozin	HPLC: C ₁₈ (100 Mm × 2.1 Mm, 3 μ m) UPLC: Hypersil Gold (50 Mm × 2.1 Mm, 1.9 μ m).	HPLC: Methanol: 0.03 M Phosphate Buffer (75:25 V/V) At Ph 3.2 UPLC: Methanol: 0.03 M Phosphate Buffer (80:20 V/V)	240 Nm	[60]
29	EMP-LIN-Tablet	ODS (250 X 4.6mm, 5µ)	Buffer: Acetonitrile (45:50 V/V)	245 Nm	[61]

Sr.	Name of drug	Sample matrix	Method	Detection (nm)	Ref.
<u>no</u> 1	LIN-pioglitazone	Synthetic	I-First derivative,	I-LIN247.80,PIO 258.40 nm;	[62]
	Ent plogituzone	mixture	II-Area under curve	II-LIN292-305, PIO265-279nm;	[02]
		mixture	III-isosbestic point	III-273.05 nm	
2	EMP-MET	Tablet	I-simultaneous equation	I-EMP224,MET233nm	[63]
2		Tablet	II-Absorption ratio	II-MET233,EMP266 nm	[00]
3	EMP-MET	Tablet	simultaneous equation	EMP-225, MET- 237 nm	[64]
4	EMP-MET	Bulk drug	absorption correction	EMP224,MET230nm	[65]
•		Dankarag		Both- 227 nm	[00]
5	EMP-LIN	Tablet	simultaneous equation	EMP233, LIN277nm	[66]
6	EMP-MET	Tablet	I- absorbance corrected	I-EMP224,MET232, intercept 203nm	[67]
-			II-area under curve	II-EMP219-229,MET 227-238nm	[]
			III- dual wavelength	III-EMP224-238,MET 232-244nm	
7	EMP-LIN	Tablet	simultaneous equation	EMP233, LIN277nm	[68]
8	EMP-MET	Tablet	I-simultaneous equation	I-EMP-272, MET- 234	[69]
-			II- Absorbance ratio	II-254nm and 226nm	[]
9	MET	Tablet	Zero order	234	[70]
10	EMP-LIN	Tablet	I-zero-order	I-EMP225.4,LIN295	[71]
			II-ratio spectra	II- EMP236.8,LIN353.8	
11	MET-LIN	Tablet	I-Third derivative	I-268.7nm	[72]
			II-Derivative ratio	II-264 nm	
			III-Ratio difference	III-262 and 268 nm	
			IV-Factorized dual wavelength	IV-264 and 268nm	
12	EMP-LIN	Tablet	I-Ratio subtraction coupled with extended ratio subtraction	I-EMP225,LIN226 nm	[73]
13	EMP	Tablet	Zero order	247	[74]

Table 2. Spectrophotometric methods used for determination of MET, EMP and LIN alone and in combined dosage form

Sr. no	Name of drug/formulation	Stationary phase plates	Mobile phase composition	Detection (nm)	Rf	Ref.
1	MET-LIN-Tablet	Silica gel 60 F254	Acetone: methanol: chloroform: formic acid (3:1:5:1v/v)	230	MET-0.19, LIN- 0.72	[75]
2	EMP-LIN-Tablet	Silica gel 60 F254	Methanol: toluene: ethyl acetate (2: 4: 4v/v/v)	229	EMP- 0.57, LIN- 0.22	[76]
3	MET-EMP-Tablet	Silica gel 60F254	Toluene: 3% ammonium acetate in methanol: ethyl acetate: ammonia (3: 5: 2: $0.4 \% v/v/v/v$)	230	MET- 0.28, EMP- 0.58	[77]
4	EMP-MET-LIN-Synthetic mixture	Silica gel 60 F254	n-Butanol: water: glacial acetic acid (6:3:1 v/v)	223	EMP-0.73, LIN- 0.52, MET-0.33	[78]
5	LIN-MET saxagliptin- vildagliptin- Synthetic mixture	silica gel 60 F254	methanol-0.5% w/v aqueous ammonium sulfate (8 : 2, v/v)	225, 208 nm	MET-0.22 LIN-0.44 - SGP-0.51 VGP- 0.46	[79]
6	MET-LIN-Tablet	silica gel 60 GF254	Acetone-methanol toluene-formic acid 4:3:2:1 (v/v/v/v)	259	MET- 0.61 LIN- 0.82	[80]

Table 4. Stability-indicating HPLC and UPLC methods for determination of MET, EMP and LIN

Sr. no	Name Of Drug/ Formulation	Stationary Phase Plates	Mobile Phase	Detection Wavelength (Nm)	Ref.
1	LIN	Zorbax SB-Aq 250 × 4.6 Mm, 5 μm	A: (0.02M) KH ₂ PO ₄ Buffer Ph 3.0; B: ACN: Meoh (90:10 V/V)	225 Nm	[81]
2	MET-EMP-Tablet	C ₁₈ BEH(Ethylene Bridged Hybrid) UPLC (100mm X 2.1mm ,1.8µm)	0.1% Ortho Phosphoric Acid Buffer (Ph 3.4): Methanol (40:60% V/V)	254 Nm	[82]
3	EMP-LIN-Tablet	Agilent C ₁₈ (4.6×150mm)5µ,	Methanol: Phosphate Buffer (KH2PO4 And K2HPO4) Phosphate Ph 3(70:30 V/V)	254 Nm	[83]
4	MET-EMP-Tablet	Kromosil C ₁₈ (250x4.6 Mm; 5.6 μ)	Acetonitrile: 0.1% Orthophosphoric Acid Buffer Ph 2.8 (50:50 V/V)	260 Nm	[84]
5	MET-EMP-Tablet	Kromasil C ₁₈ (250mm×4.6mm, 5 μm)	Buffer: Acetonitrile (45: 55 V/V)	226nm	[85]

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Sr. no	Name Of Drug/ Formulation	Stationary Phase Plates	Mobile Phase	Detection Wavelength (Nm)	Ref.
6	EMP-Tablet	Develosil ODS HG-5 RP C ₁₈ , 5µm, 15cm X 4.6mm	Phosphate Buffer Ph-2.8: Acetonitrile (45:55 V/V)	228 Nm	[86]
7	LIN-Tablet	C18 Column	Acetonitrile: Methanol (50:50 (V/V)	238 Nm	[87]
8	MET-EMP-Synthetic Mixture	Kromosil 250 X 4.6 Mm, 5µm	0.1% Ortho Phosphoric Acid Buffer: Acetonitrile (45:55v/V)	233 Nm	[88]
9	MET-LIN-Tablet	BDS Hypersil C ₈ (250 × 4.6 Mm, 5 μ m)	Acetonitrile: Water: Methanol (25:50:25 V/V/V)	243 Nm	[89]
10	EMP-Tablet	Intersil® C ₁₈ (150 Mm × 4 Mm, 5 Mm)	Acetonitrile: Phosphate Buffer Ph 4, (50:50 V/V)	225 Nm	[90]
11	EMP-LIN-Tablet	BDŚ C ₁₈ (250mm X 4.6 Mm 5µ)	0.1% Perchloric Acid: Acetonitrile (60:40 V/V)	230nm	[91]
12	EMP-MET-Tablet	Inertsil ODS (4.6 X 150mm, 5 μ M)	Buffer (Ph 3): Methanol (30:70 V/V)	220 Nm	[92]
13	EMP-LIN-Tablet	Luna Omega Polar C ₁₈ , 100x2.1mm, 1.6µ	A: 10mm Potassium Dihydrogen Orthophosphate Ph-3.0; B: Acetonitrile And Methanol (55:45%V/V)	225nm	[93]
14	MET-LIN-Tablet	Inertsil ODS2, 150 Mm X 4.6mm, 5µ	Ammonium Phosphate Buffer (Ph 3.00): Methanol (40:60 V/V)	233nm	[94]
15	LIN-Tablet	Thermo Scientific® RP-8 (100 Mm × 4.6 Mm; 5µm)	A: 0.1% Formic Acid Ph 3.5; B:Acetonitrile	294 Nm	[95]
16	MET-EMP-Tablet	BDS (250 Mm × 4.6 Mm, 5 M)	0.1% Orthophosphoric Acid Buffer: Acetonitrile (50:50 V/V)	210 Nm	[96]
17	MET-EMP-Tablet	Thermosil C ₁₈ (4.6mmx250mm, 5 μ).	Methanol: Acetonitrile: 0.025M Potassium Hydrogen Phosphate Buffer (Ph: 3) (45:30:25, V/V/V)	225nm	[97]
18	MET-EMP-Tablet	Kromasil C ₁₈ (250mm×4.6mm, 5μm),	Buffer: Acetonitrile (45:55 V/V)	226nm	[98]
19	EMP- LIN-Tablet	Hypersil ODS 3V, 250 X 4.6 Mm.5.0µ	A: Potassium Di-Hydrogen Phosphate Ph2.20 B: Water: Acetonitrile (5:95)	225nm	[99]

Sr.	Name of drug/formulation	Stationary phase plates	Mobile phase	MS/MS transitions	Ref.
no				(m/z)	
1	MET-Alogliptin-Tablet	Hypersil gold 50 mm × 2.1 mm	Acetonitrile:0.2 % formic acid	MET-130.12-71.32,	[100]
		(1.9 µm)	(10: 90, v/v)	ALN-340.33-116.32	
2	EMP-Human plasma	UPLC BEH Shield RP C ₁₈ (150	Water: acetonitrile (10:90,	449.01- 371.21	[101]
		×2.1 mm, 1.7 μm)	v/v)		
3	EMP-MET-Tablet	C ₁₈ (50 × 2.1 mm, 1.7µm	0.1% Aqueous formic acid:	EMP-451.04–71.07, MET-130.11–	[102]
			acetonitrile (75:25, v/v)	71.14	
4	EMP	Eclipse plus C_{18} RRHD (2.1 × 50	Water with 0.1% formic acid:	Base peak-451.1516	[103]
		mm, 1.8 μm)	methanol in 50:50 (v/v)		
5	LIN- tadalafil- plasma	Zorbax Eclipse XDBC ₁₈ (4.6 ×	Methanol:0.05% aqueous	Base peak-LIN-473, TDL-390	[104]
		150 mm, 5 μm)	formic acid (50:50)		
6	LIN-MET- human plasma	Symmetry® C ₁₈ (3.5 µm x 4.6	Methanol: 10 mM ammonium	LIN-473.24-419.94 MET-130.14-	[105]
		mm x 50)	formate buffer (containing 0.2	60.18	
			% formic acid) (95: 5, v/v)		
7	EMP-MET- Canagliflozin-	Xbridge C ₁₈ column (50 × 2.1	A: water;	MET- 130.1-71.1	[106]
	Dapagliflozin-Plasma	mm, 5 μm)	B: acetonitrile containing 1	EMP-468.0-354.9 CAN-462.0-	
			mM ammonium formate and	191.2 DAP-426.1-167.1	
			0.1% formic acid.		
8	LIN-Plasma	Waters, X-Bridge, C ₁₈ , (5µm	Acetonitrile: 0.1 % formic acid	473.54-157.6	[107]
		4.6×50 mm)	(90:10 v/v)		
9	EMP-MET-Plasma	Orosil C ₁₈ column (150 4.6 mm,	Methanol:10 mM ammonium	EMP-468.0-355.1	[108]
		3 m).	trifluoroacetate (90:10, v/v)	MET-130.0-71.2	
10	LIN-Plasma	Gemini C ₁₈ (100 mm × 4.6 mm,	10 mM Ammonium formate:	473.3-420.1	[109]
		3µ)	methanol (20:80 v/v)		
11	EMP-LIN-Plasma	XSelect HSS Cyano (50 ×2.1	A: 2Mm ammonium acetate,	EMP- 451.3-71.1	[110]
		mm,3.5μm)	pH 5.0	LIN- 473.2-420.2	
		· · /	B: acetonitrile		

Table 5. LC-MS/MS method for determination of MET, EMP and LIN

3. CONCLUSION

There are a extensive choice of analytical techniques were accessible for the analysis of MET. EMP and LIN alone or in combination with other drugs in pharmaceutical and biological samples. The available data it was revealed that HPLC technique was comprehensively used for the estimation of MET, EMP and LIN in various matrices like plasma and serum. HPLC MS/MS also recommended method in is the determination of MET, EMP and LIN in biological samples because HPLC separation ability with MS sensitivity and selectivity allows the unambiguous identification of MET, EMP and LIN and its metabolites. HPLC with UV detection is applicable in the case of analysis of MET, EMP and LIN in pharmaceuticals which provide us cost effective accurate method when compare more advance with techniques. Various bioanalytical method development and validation has been performed in order to study the bioavailability and bioequivalence of the drug which would help the chemists/analyses to formulate a stabilized drug.

DISCLAIMER

The products used for this research are commonly and predominantly use products in our area of research and country. There is absolutely no conflict of interest between the authors and producers of the products because we do not intend to use these products as an avenue for any litigation but for the advancement of knowledge. Also, the research was not funded by the producing company rather it was funded by personal efforts of the authors.

CONSENT

It is not applicable.

ETHICAL APPROVAL

It is not applicable.

COMPETING INTERESTS

Authors have declared that no competing interests exist.

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